

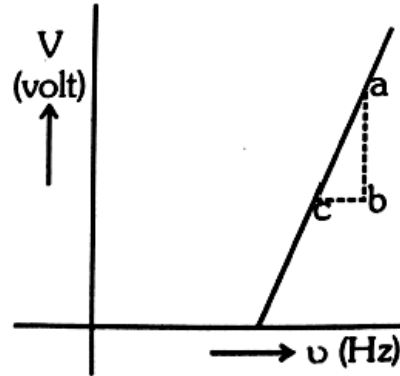
# 32<sup>nd</sup> Session: Modern Physics III

## Classification of Nuclei

- Recap
- 2<sup>nd</sup> classification of nuclei
  - Binding energy & mass defect
  - Stable & unstable (**radioactive**) nuclei (**2 important curves**)
    - N v/s P
    - B.E. per Nucleon v/s A

Q. 1). In a photoelectric experiment the graph of frequency  $\nu$  of incident light (in Hz) and stopping potential  $V$  (in volt) is shown in below. From figure the value of the Planck's constant is ( $e$  is the elementary charge)

- (1)  $e \frac{ab}{cb}$       (2)  $e \frac{cb}{ab}$   
 (3)  $e \frac{ac}{bc}$       (4)  $e \frac{ac}{ab}$



$$eV_s = h\nu - \phi$$

$$V_s = \frac{h}{e}\nu - \frac{\phi}{e}$$

$$\frac{h}{e} = \tan\theta = \text{slope}$$

$$= \frac{ab}{cb}$$

$$h = e \cdot \frac{ab}{cb}$$

Q. 2).

A 500 W bulb is placed at the center of a perfectly black sphere of radius  $R = 1$  metre. The approximate pressure experienced by the walls of the sphere as it absorbs all the photons emitted by the bulb is (take  $4\pi = 12.6$ )

(1)  $1.3 \times 10^{-7} \text{ N/m}^2$  (2)  $2.5 \times 10^{-7} \text{ N/m}^2$

(3)  $6.3 \times 10^{-7} \text{ N/m}^2$  (3)  $3.14 \times 10^{-7} \text{ N/m}^2$



$$N_p = \frac{500}{4\pi(1)^2} \times \frac{SA}{hc} \lambda$$

$$\frac{\Delta P}{\Delta t} = \frac{h}{\lambda \Delta t}$$

press = force/area

$$= \frac{500}{4\pi c} = \frac{500}{12.6 \times 3 \times 10^8} \quad \text{force} = N_p \times \frac{h}{\lambda}$$

$$= 1.3 \times 10^{-7} \text{ N/m}^2 \quad = \frac{500}{4\pi} \times \frac{SA}{hc} \times \frac{h}{\lambda}$$

Q. 3). A laser beam ( $\lambda = 633 \text{ nm}$ ) has an power of 3 mW. What will be the pressure exerted on a surface by this beam if the cross sectional area is  $3 \text{ mm}^2$ . (Assume perfect reflection and normal incidence)

- (1)  $6.6 \times 10^{-3} \text{ N/m}^2$       ✓ (2)  $6.6 \times 10^{-6} \text{ N/m}^2$   
 (3)  $6.6 \times 10^{-9} \text{ N/m}^2$       (4)  $6.6 \text{ N/m}^2$

$$\text{pressure} = \frac{3 \times 10^{-3}}{hc} \times \frac{2h}{\lambda} \times \frac{1}{3 \times 10^{-6}} = 6.6 \times 10^{-6} \text{ N/m}^2$$

- Steps (I AM)
- 1) Find  $(N_p)$   $\rightarrow$  normal area
  - 2)  $\Delta p$  for each photon
  - 3) force =  $N_p \times \Delta p$   
 $\downarrow$   
 components or not.
  4. vector add<sup>n</sup>
  - 5). press = force / normal area.

Q. 4). The electron of a hydrogen atom revolves round the proton in a circular  $n^{\text{th}}$  orbit of radius  $r_n = \frac{\epsilon_0 n^2 h^2}{\pi m e^2}$  with a speed  $v_n = \frac{e^2}{2\epsilon_0 n h}$ . The current due to the circulating charge is proportional to

(1)  $e^2$

(2)  $e^3$

~~(3)  $e^5$~~

(4)  $e^6$

$$i = \frac{e}{T} = \frac{e v_n}{2\pi r_n} \propto \frac{e \cdot e^2 \times e^2}{1} \propto e^5$$

Q. 5). Energy levels A, B and C of a certain atom correspond to increasing values of energy i.e.  $E_A < E_B < E_C$ . If  $\lambda_1$ ,  $\lambda_2$ , and  $\lambda_3$  are wave lengths of radiations corresponding to transitions C to B, B to A and C to A respectively, which of the following relations is correct :-

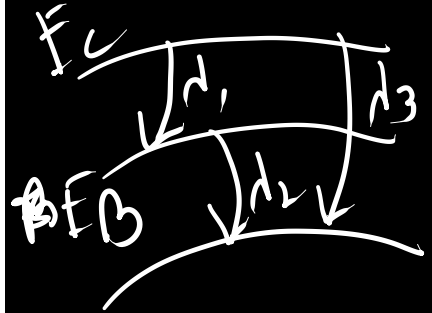
(1)  $\lambda_3 = \lambda_1 + \lambda_2$

(2)  $\lambda_3 = \frac{\lambda_1 \lambda_2}{\lambda_1 + \lambda_2}$

(3)  $\lambda_1 + \lambda_2 + \lambda_3 = 0$

(4)  $\lambda_3^2 = \lambda_1^2 + \lambda_2^2$

$\lambda_3 = \frac{\lambda_1 \lambda_2}{\lambda_1 + \lambda_2}$



$\frac{hc}{\lambda_3} = \frac{hc}{\lambda_1} + \frac{hc}{\lambda_2}$

$\frac{1}{\lambda_3} = \frac{1}{\lambda_1} + \frac{1}{\lambda_2}$

$\lambda_3 = \frac{\lambda_1 \lambda_2}{\lambda_1 + \lambda_2}$

Q. 6). Electrons of mass  $m$  with de-Broglie wavelength  $\lambda$  fall on the target in an X-ray tube. The cutoff wavelength ( $\lambda_0$ ) of the emitted X-ray is :-

(A)  $\lambda_0 = \frac{2m^2c^2\lambda^3}{h^2}$

(2)  $\lambda_0 = \lambda$

(3)  $\lambda_0 = \frac{2mc\lambda^2}{h}$

(4)  $\lambda_0 = \frac{2h}{mc}$

$$d_e = \frac{h}{p} = \frac{h}{\sqrt{2mk}} \Rightarrow k = \frac{h^2}{2m d_e^2} = \frac{hc}{\lambda_0}$$

$$\lambda_0 = \frac{hc \cdot 2m d_e^2}{h^2} = \frac{2mc d_e^2}{h}$$

Q. 7). For production of characteristics  $K_{\beta}$  X-rays, the electron transition will be :-

(1)  $n = 2$  to  $n = 1$

(2)  $n = 3$  to  $n = 2$

(3)  $n = 3$  to  $n = 1$

(4)  $n = 4$  to  $n = 2$



Q. 8) if voltage of X-ray tube is doubled then intensity of X-rays will :-

- (1) Halved                      ✓ ~~(2) Remains constant~~  
(3) Doubled                      (4) Quadrupled

Q. 9). 20 kV potential is applied across X-ray tube, the minimum wavelength of X-ray emitted will be :-

✓ (1) 0.62 Å

(2) 0.37 Å

(3) 1.62 Å

(4) 1.31 Å

$$20 \text{ kV}_e = \frac{hc}{\lambda}$$
$$\lambda = \frac{12400}{20 \times 1000} = 0.62 \text{ Å}$$

Q. 10). Lattice constant of a crystal is  $3 \times 10^{-8}$  cm and glancing angle of X-ray is  $30^\circ$  for first order diffraction, then the value of  $\lambda$  will be :-

- (1)  $6 \times 10^{-8}$  cm                       (2)  $3 \times 10^{-8}$  cm  
 (3)  $1.5 \times 10^{-8}$  cm                      (4)  $10^{-8}$  cm

Lattice constant  $\leftrightarrow$  inter-planar dist.

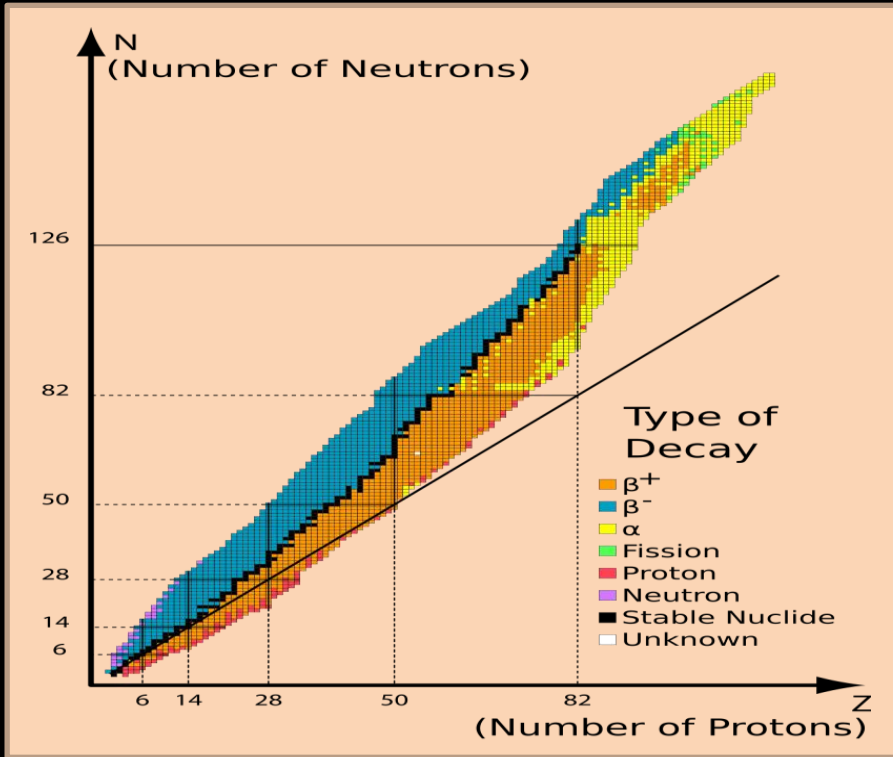
$$2d \sin \theta = n\lambda$$

$$\Rightarrow 3 \times 10^{-8} \sin 30^\circ = 1 \times \lambda$$

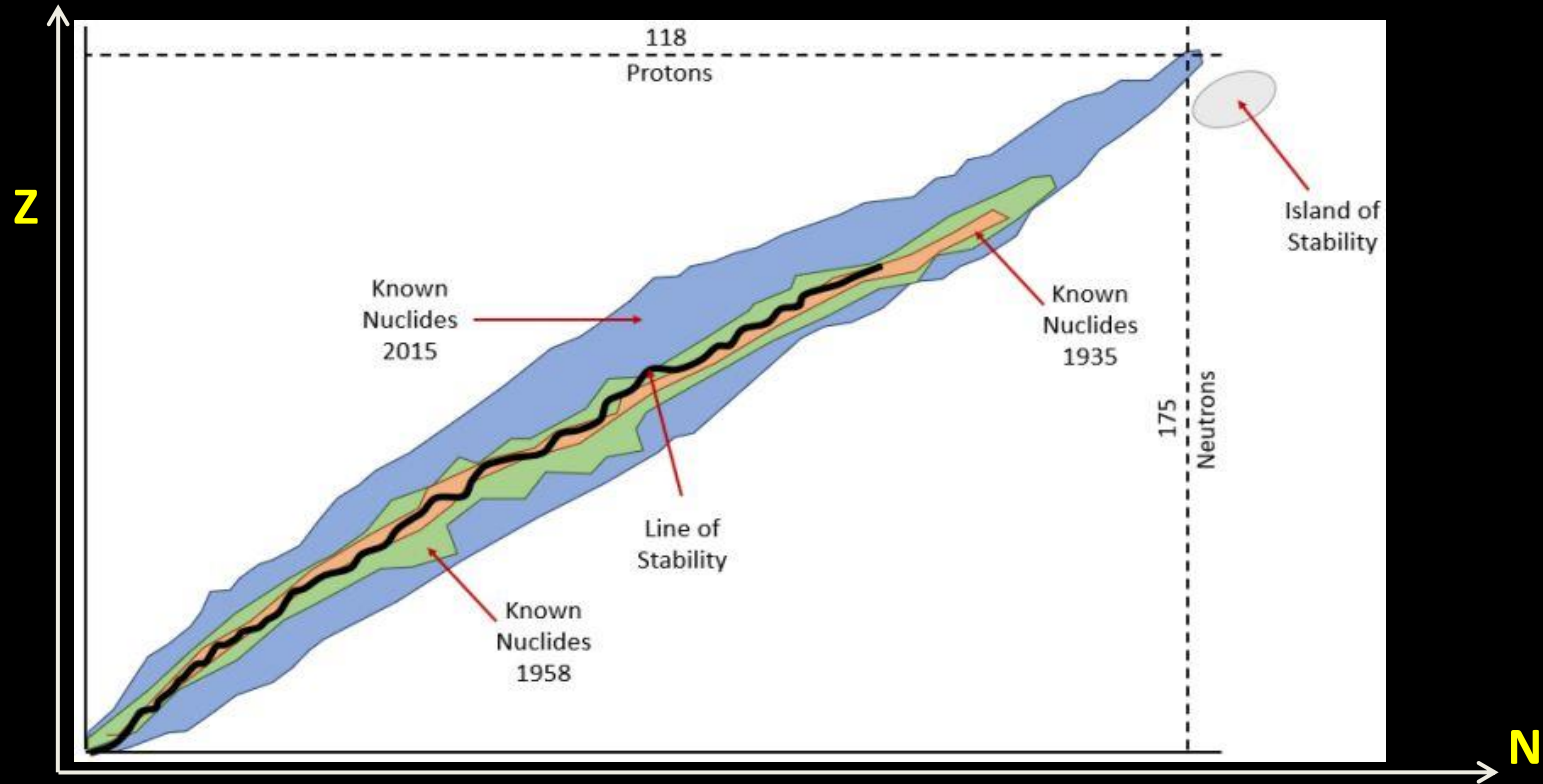
$$\lambda = 3 \times 10^{-8} \text{ cm}$$

<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>	<b>8</b>	<b>9</b>	<b>10</b>
<b>1</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>2</b>	<b>3</b>	<b>3</b>	<b>2</b>	<b>1</b>	<b>2</b>

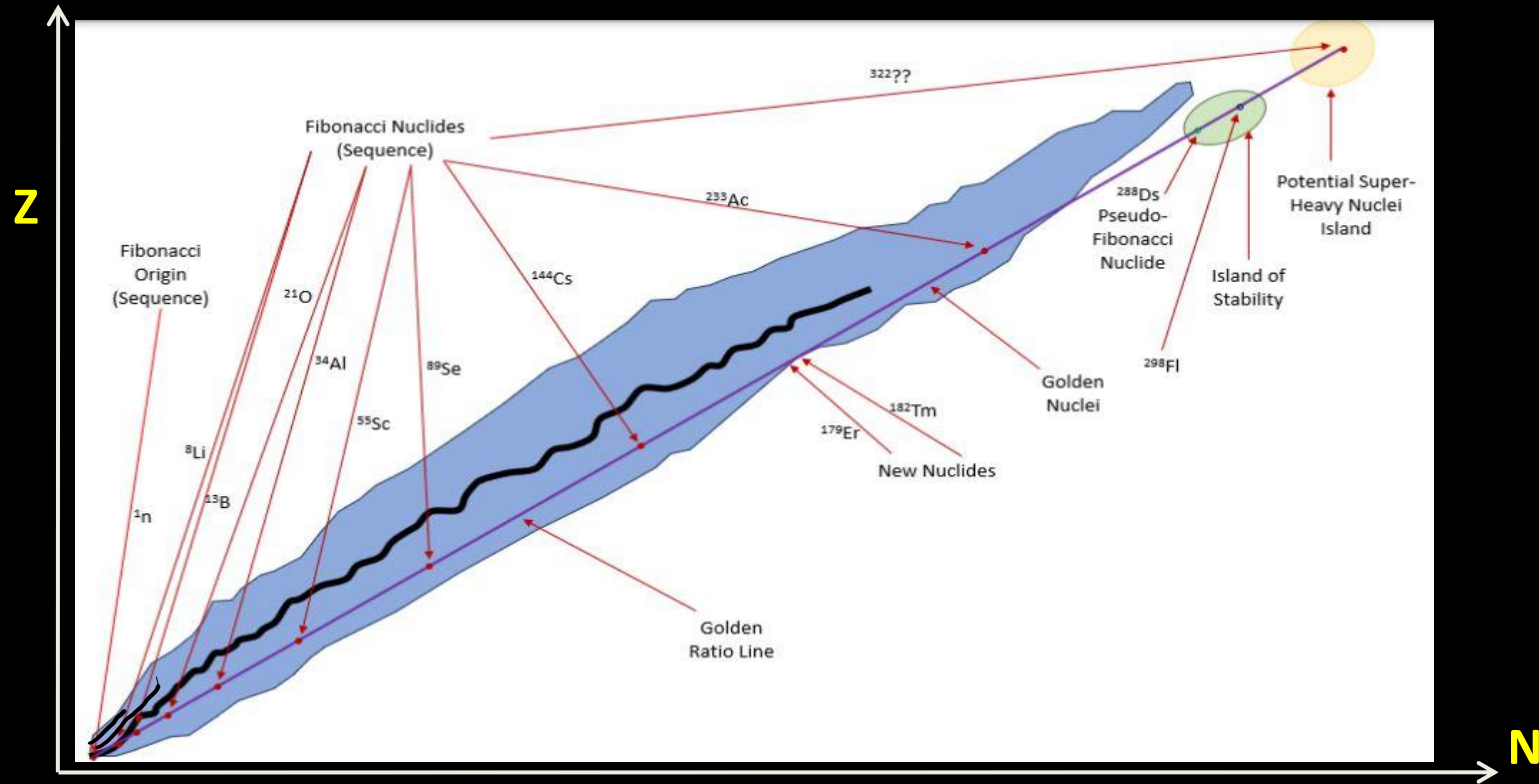
# 2<sup>nd</sup> Classification (based on stability): Chart of nuclides

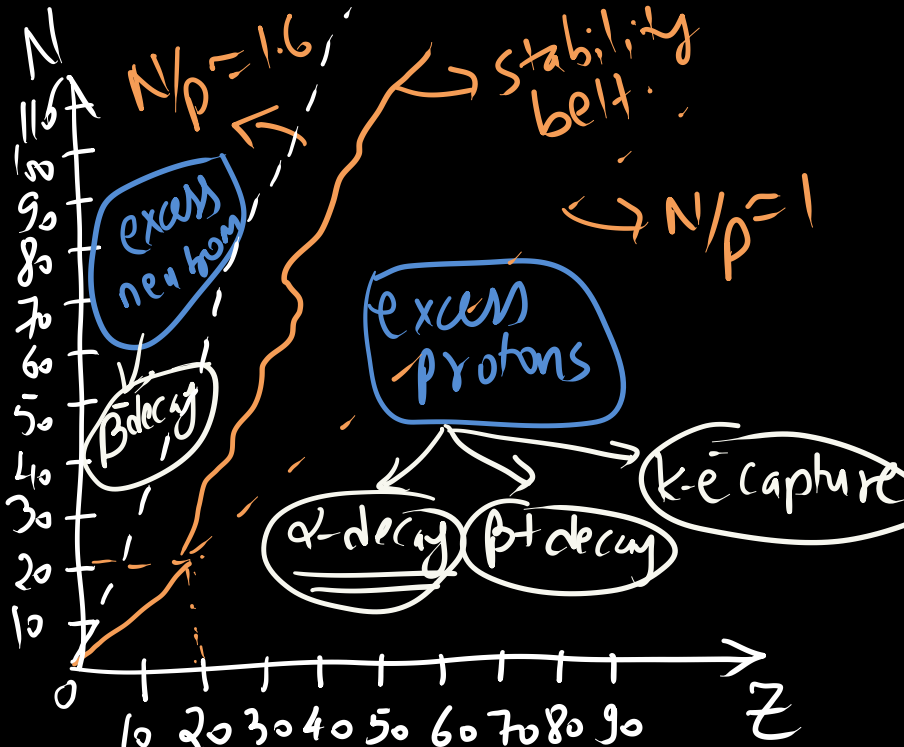


# Chart of Nuclides



# Chart of Nuclides: Golden Ratio line





① For  $Z < 20$

stable nuclei have  $n/p \approx 1$

$4\text{He}$ 2	$12\text{C}$ 6	$14\text{N}$ 7	$16\text{O}$ 8	$23\text{Na}$ 11	$24\text{Mg}$ 12	$40\text{Ca}$ 20
$p \rightarrow 2$	$p \rightarrow 6$	$p \rightarrow 7$	$p \rightarrow 8$	$p \rightarrow 11$	$p \rightarrow 12$	$p \rightarrow 20$
$n \rightarrow 2$	$n \rightarrow 6$	$n \rightarrow 7$	$n \rightarrow 8$	$n \rightarrow 12$	$n \rightarrow 12$	$n \rightarrow 20$
$1\text{H}$ 1	$3\text{He}$ 2	$9\text{Be}$ 4	$7\text{Li}$ 3	$12/11 = 1.09$		
$p \rightarrow 1$	$p \rightarrow 2$	$p \rightarrow 4$	$p \rightarrow 3$			
$n \rightarrow 0$	$n \rightarrow 1$	$n \rightarrow 5$	$n \rightarrow 4$			
	$n/p = 5/4 = 1.25$	$n/p = 4/3 = 1.33$				

③ An unstable nucleus disintegrates to become stable by changing its  $(n/p)$  ratio.

② for  $Z > 20$

stable nuclei have  $1 < n/p < 1.618$

$56\text{Fe}$ 26	$p \rightarrow 26$	$n \rightarrow 30$	$88\text{Ag}$ 47	$p \rightarrow 47$	$n \rightarrow 61$	last stable nucleus $\rightarrow 209\text{Bi}$ 83
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