

Modern Physics, Session 3: 3<sup>rd</sup> Sept 2022, Saturday, 6.00 – 7.15 pm

Q 1. An electromagnetic wave in vacuum:  $E = E_0 \sin(kx - \omega t)$ . Which is independent of wavelength?

- Today's subtopics on dual nature of light:
- 1). Recap
  - 2). Time delay & frequency perspectives
  - 3). **Photon's Birthday**
  - 4). Photon vs material particles
  - 5). Photoelectric effect setup

- A).  $k$
- B).  $\omega$
- C).  $k/\omega$
- D).  $k\omega$

$k = \text{no. of waves per unit m}$

$k = 2\pi/\lambda$        $\omega = 2\pi f$        $\omega = 2\pi \frac{c}{\lambda}$        $c = f\lambda$

$k/\omega = \frac{2\pi}{\lambda} \cdot \frac{1}{2\pi f} = \frac{2\pi}{\lambda} \cdot \frac{\lambda}{2\pi c} = 1/c$

$v = f\lambda$       Mnemonic

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Q 2. A free electron is placed in the path of a plane electromagnetic wave. The electron will start moving

- A). along the electric field
- B). along the magnetic field
- C). along wave propagation direction
- D). in a plane containing magnetic field & direction of propagation

Q 3). Speed of EM waves is the same

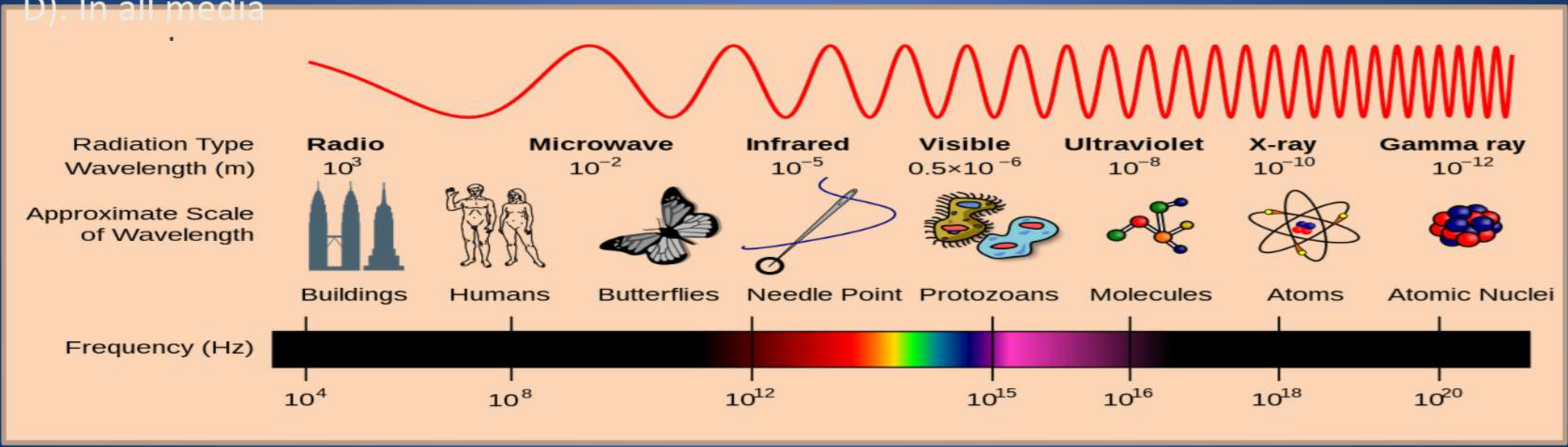
- A). for all wavelengths
- B). for all intensities
- C). in all inertial frame
- D). In all media

Maxwell derived speed of EM wave,  $v = E_0/B_0$

Constant.  $v = \frac{1}{\sqrt{\mu_0 \epsilon_0}} = 3 \times 10^8 \approx 299,792,458 \text{ m/s}$

$\leftarrow$   $\leftarrow$   $\rightarrow$

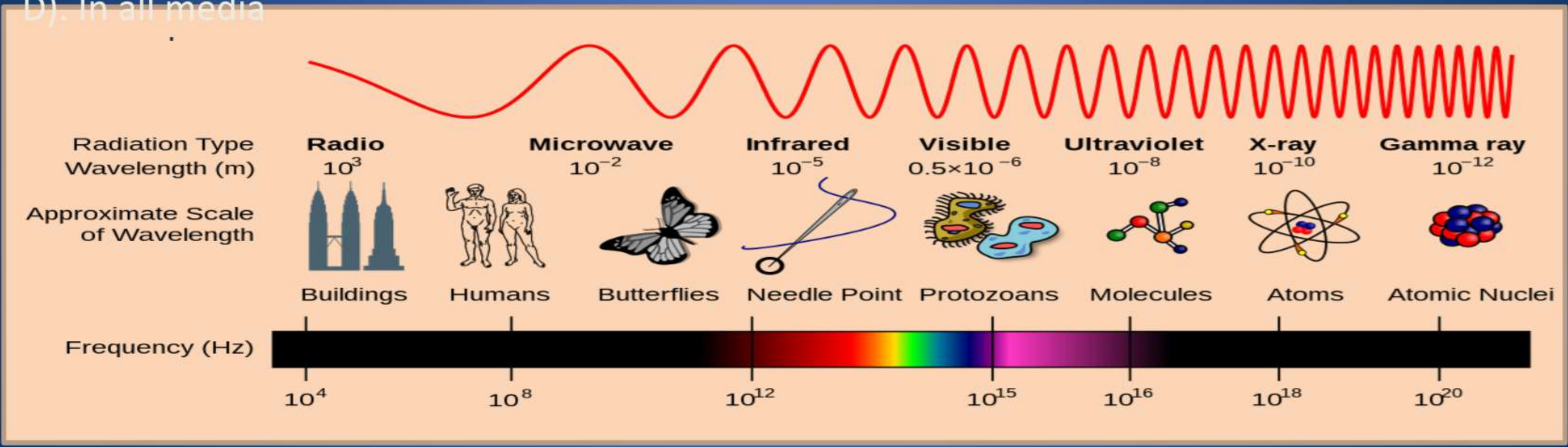
$4\pi \times 10^{-7}$



No mechanics experiment  
can tell if the for is moving  
or at rest.

Q 3). Speed of EM waves is the same

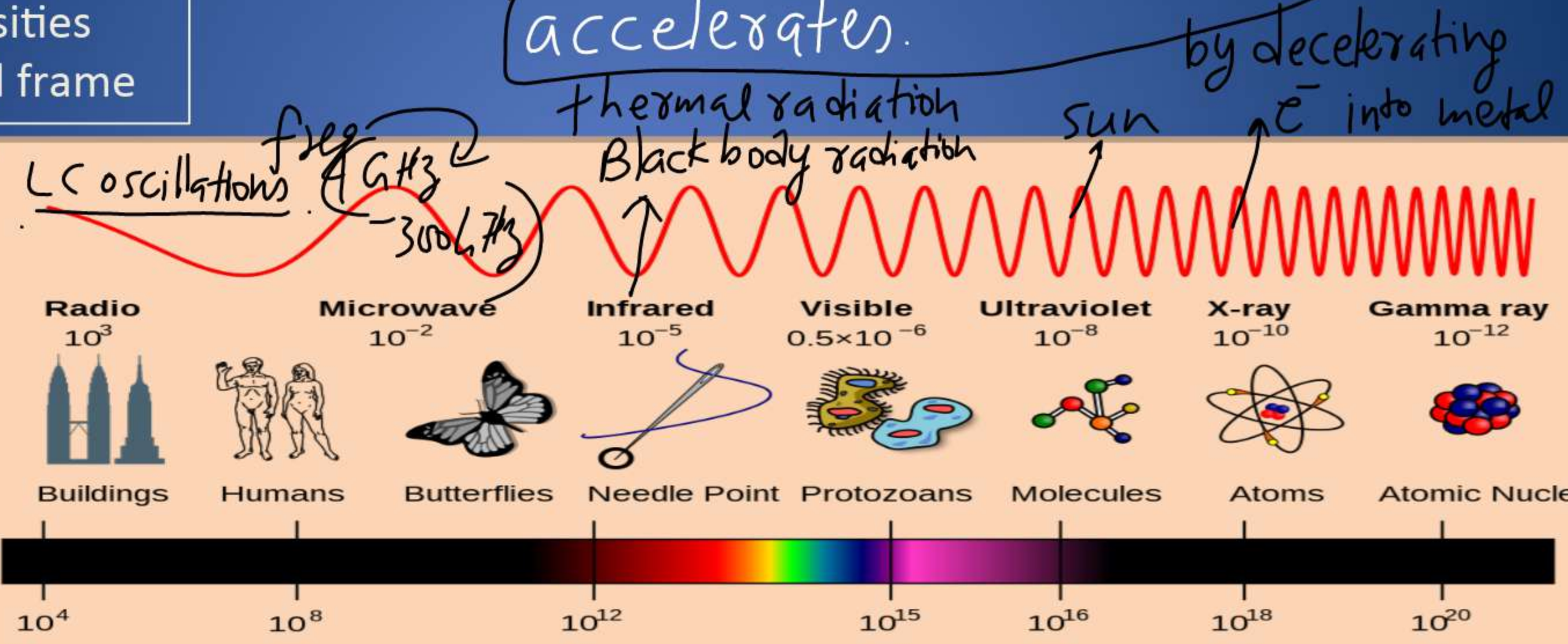
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Maxwell said that an EM wave is generated whenever a charged particle accelerates.



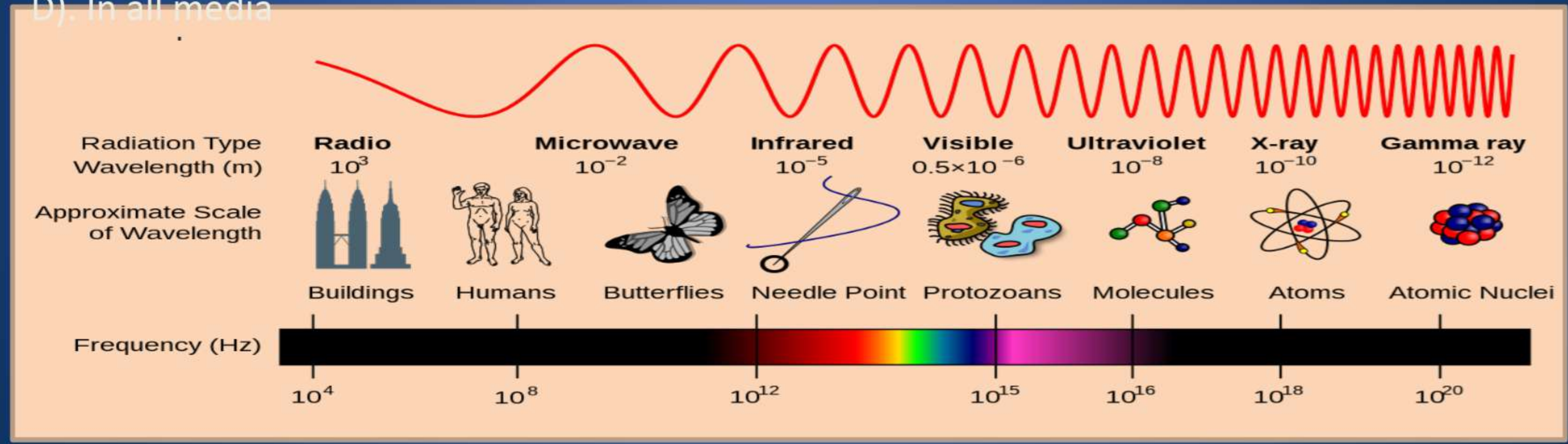
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Q 4). Which have zero average value in a plane EM wave

$$E = E_0 \sin(\omega t - kx)$$

- A). Electric Field
- B). Magnetic Field
- C). Electric Energy
- D). Magnetic Energy

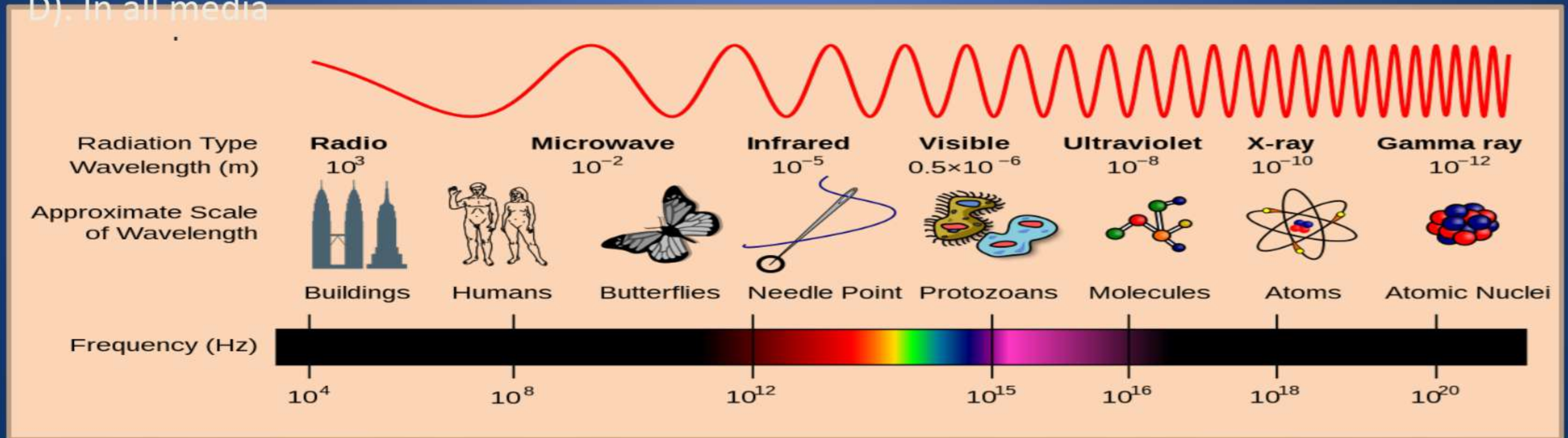


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- ✓ A). Electric Field
- ✓ B). Magnetic Field
- C). Electric Energy
- D). Magnetic Energy



Q 5). Intensity of sound wave is proportional to

- A). (amplitude)<sup>2</sup> ✓
- B). (frequency)<sup>2</sup> →
- C). density
- D). wave number

$$\propto \frac{\omega}{2\pi f}$$

$$\begin{aligned}
 I &= \frac{1}{2} \rho v \omega^2 s_0^2 \\
 &= \frac{\rho^2}{2\rho v} \\
 &= \frac{1}{2} \rho \omega^2 s_0^2 \rightarrow \omega
 \end{aligned}$$



Q 5). Intensity of sound wave is proportional to

- A). (amplitude)<sup>2</sup>
- B). (frequency)<sup>2</sup>
- C). density
- D). wave number

Q 6). Intensity of light/EM wave is proportional to

- A). (amplitude)<sup>2</sup>
- B). (frequency)<sup>2</sup>
- C). density
- D). wave number

Today's session starts now...

Sh-> Why is the ejection of electrons from the metal surface, a threat to wave nature of light?

G -> Light is an EM wave was just confirmed.  
And then

G-> How can you produce light?

- A). Heating a filament
- B). Friction
- C). Moving electrons
- D). Accelerating electrons

Hertz, Hallwachs & Lenard in 1887-88, observed the spark in an attempt to produce EM wave by accelerating the electrons produced.

G-> The spark is **barely visible** with enormous visible light but became vigorous upon **small UV** light exposure.

G-> Three perspectives to the threat to wave nature.

- 1). **Intensity perspective**
- 2). Time delay perspective
- 3). Frequency perspective

Today's session starts now...

Sh-> Why is the ejection of electrons from the metal surface, a threat to wave nature of light?

Intensity Perspective →

$$I = \frac{1}{2} \epsilon_0 E_0^2 c$$

Total energy incident. ←  $U = I A t$   
 $\phi$ , wave  $f$

force on a  $e^-$  →  $e E_0$   
 (free  $e^-$  in the metal)

G-> Three perspectives to the threat to wave nature.

- 1). Intensity perspective
- 2). Time delay perspective
- 3). Frequency perspective

Sh -> What is Time Delay Perspective ?

G-> It is the **minimum time** that **light wave** would take to cause emission  
Solve the following:

$\bar{e}$  should take some time to come out of metal surface.

Sh -> What is Time Delay Perspective ?

G-> It is the **minimum time** that **light wave** would take to cause emission  
Solve the following:

G-> P = 1.0 Watt. Potassium foil at R = 0.5 m. Assume wave nature of light beam with radius equal to the radius of K atom, r = 1.3 Å. Find time taken for e<sup>-</sup> to come out. (φ<sub>K</sub> = 2.2 eV)

- A). 20.7 s
- B). 1 μs
- C). 10 s
- D). 1 hour

$P \rightarrow \text{Watt} \rightarrow \text{J/s}$   
 $R = 0.5 \text{ m}$   
 $r = 1.3 \text{ \AA} = 1.3 \times 10^{-10} \text{ m}$   
 $\phi = 2.2 \text{ eV}$

$$I = \frac{P}{A} = \frac{1.0}{4\pi R^2} = \frac{1}{4\pi (0.5)^2} \text{ W/m}^2$$

$$U = I \times \pi r^2 \times t_{\text{min}} = \phi = 3.52 \text{ eV}$$

$$t_{\text{min}} = \frac{2.2 \times 1.6 \times 10^{-19} \times 4\pi (0.5)^2}{1 \times \pi \times (1.3 \times 10^{-10})^2} = \frac{2.2 \times 1.6 \times 10^{-19} \times 2\pi}{\pi \times 1.69} = \frac{3.52}{1.69} \text{ s} \approx 20.7 \text{ s}$$

Sh-> How is wave and particle nature related?

G-> Light particle is called as **photon**.  
Its **energy** is related through the **frequency** of light wave  
Each photon is an energy packet

$$E_{\text{photon}} \propto f_{\text{light wave}}$$